## Novel chlorin-based red-light responsive dual-effective reagents for PDT: Synthesis, DNA-binding and pre-cytotoxicity studies

Ali Tuna, Petr Rathner, Günther Knör

Institute of Inorganic Chemistry, Johannes Kepler University, 4040 Linz, Austria

Chlorins are  $18\pi$ -electron conjugated aromatic tetrapyrrole pigments with chlorophylls being their most prominent representatives [1]. In contrast to porphyrins, in chlorin-type ligands one of the double bonds of the tetrapyrrole macrocycle is hydrogenated. This strongly enhances their possibility to absorb red light, which is a crucial design principle for biomedical applications in living systems such as photodynamic therapy (PDT) and photopharmacological processes [2].

In previous studies, already some platinum complex decorated porphyrin systems were designed as potentially dual-effective anticancer drugs and their additive cytotoxic and photodynamic properties were characterized to a certain extent [3,4].



In the present study, we employed better synthetic approaches on the preperation of novel red-light responsive and dual-effective metallochlorin derivatives. An in-depth photochemical characterization, as well as first fundamental DNA-binding and cytotoxicity tests are currently starting, ultimately aiming at better understanding of the physiological activity of such compounds in living cells controlled under light exposure.

<sup>[1]</sup> Ester Borbas, Handbook of Porphyrin Science 2016, volume 36, chapter 181 – Chlorins.

<sup>[2]</sup> Knör, G. et. al., ChemPhotoChem 2017, 1, 378.

<sup>[3]</sup> Brunner, H. et. al, Angew. Chem., Int. Ed. 1994, 33, 2214.

<sup>[4]</sup> Spingler, B. et al., Angew. Chem., Int. Ed. 2014, 53, 6938; Chemistry Eur. J. 2015, 21, 1179.