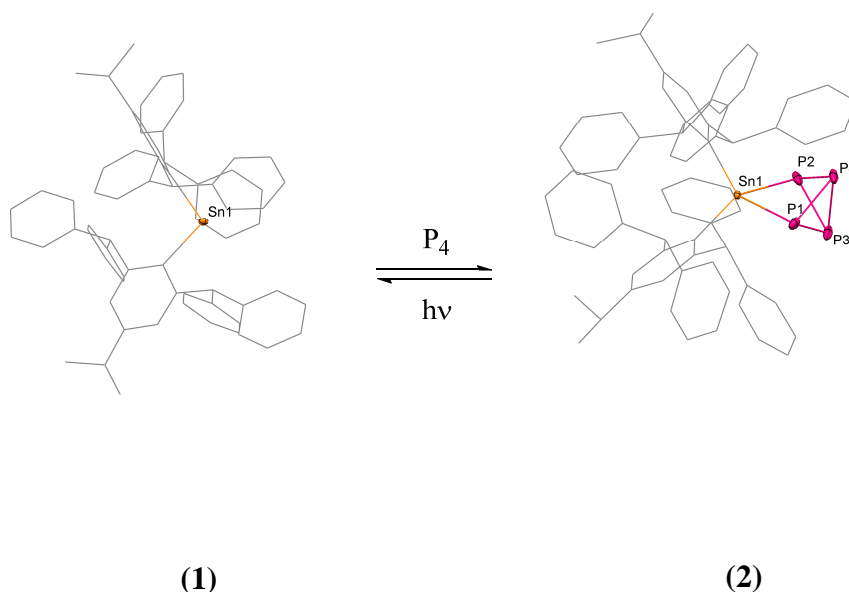


# An air-stable storage compound for white Phosphorus: Reversible Addition of P<sub>4</sub> to a Stannylene

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Despite its toxic and pyrophoric nature, the application of white phosphorus as a precursor for phosphorus containing compounds is inevitable. [1] Over the last few decades impressive progress was achieved in the targeted conversion of this highly reactive compound using transition metals and main group element compounds over last few decades. [2, 3] In the case of group 14 compounds, reaction chemistry is dominated by NHCs, CAACs and silylenes and examples for heavier group 14 homologues are scarce. [4, 5] Herein, we present the controlled insertion of a stannylene **1** into a single P-P bond of white phosphorus to give the non-pyrophoric, air-stable compound **2**. Upon irradiation with light (354 nm) the title compound releases P<sub>4</sub> quantitatively. Calculations, kinetic data and follow-up chemistry will be presented.



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